

## INDIAN INSTITUTE OF TECHNOLOGY TIRUPATI

## भारतीय प्रौद्योगिकी संस्थान तिरुपति

1.	Title of the course	Advanced Thermodynamics
2.	Course number	ME601L
3.	Structure of credits	3-0-0-3
4.	Offered to	PG
5.	New course/modification to	Modification To ME6021/20
6.	To be offered by	Department of Mechanical Engineering
7.	To take effect from	January 2022
8.	Prerequisite	СоТ
9.	Course Objective(s): To discuss thermodynamic principles and their relevance to the practical problems. To apply thermodynamic principles to solve wide range of engineering problems.	
	Course Content: Recapitulation of fundamentals: Laws of thermodynamics, Entropy and concept of irreversibility, mechanism of entropy generation; Second law analysis of closed and open systems; Exergy analysis: concepts, exergy balance, exergy transfer, exergetic efficiency, exergy analysis of power and refrigeration cycles; Single and multi-phase systems: Maxwell's relations, Clausius-Clapeyron equation, phase diagrams, equation of state; Chemically reacting system: chemical reactions, irreversible reactions; Entropy generation minimization and application to practical problems; Kinetic theory of gasses.	
10.	of irreversibility, mechanism of entropy general systems; Exergy analysis: concepts, exergy be analysis of power and refrigeration cycles; Si Clausius-Clapeyron equation, phase diagram chemical reactions, irreversible reactions; En	eration; Second law analysis of closed and open alance, exergy transfer, exergetic efficiency, exergy ngle and multi-phase systems: Maxwell's relations, s, equation of state; Chemically reacting system:
11.	of irreversibility, mechanism of entropy general systems; Exergy analysis: concepts, exergy be analysis of power and refrigeration cycles; Si Clausius-Clapeyron equation, phase diagram chemical reactions, irreversible reactions; Expractical problems; Kinetic theory of gasses.  Textbook(s):  1. Bejan A, Advanced Engineering Thermodyna	eration; Second law analysis of closed and open alance, exergy transfer, exergetic efficiency, exergy ngle and multi-phase systems: Maxwell's relations, s, equation of state; Chemically reacting system: ntropy generation minimization and application to